

Loose-Fit Polyrotaxane Composed of γ -Cyclodextrin and Single Poly(Ethyelene Glycol) Chain: Making Room in γ -CD Cavity for Additional Inclusion Complexation

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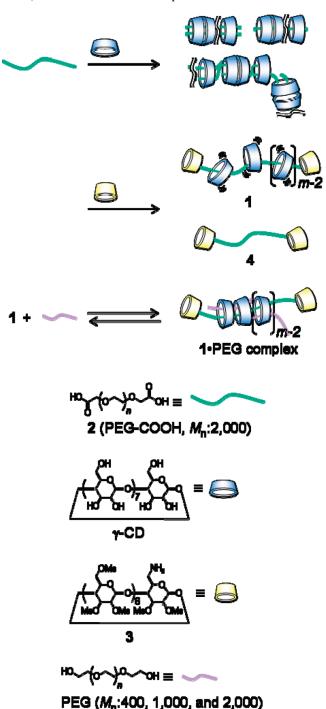
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Cyclodextrin (CD)-based inclusion complexes have been paid much attention in the last decades for their potential as supermolecules, 1,2 in which CD molecules are assembled onto the linear polymeric guest molecules with a suitable size so as to fill the cavity.³ It is worth mentioning that γ -CD has been reported to include two chains of linear polymeric guest such as poly(ethylene glycol) (PEG), although α-CD forms a single-stranded inclusion complex with such a polymeric chain. Because it is easily imagined that a structure of single-stranded inclusion complexes is unambiguous, they are important precursors for the endcapping reaction at the terminals of polymeric chain with bulky molecules to give polyrotaxanes.^{1,5} Also, the terminals of a polymeric chain in the inclusion complex of γ -CD would undergo an end-capping reaction, even with its ambiguous structures. We envisaged that a "loose" γ -CD-based polyrotaxane 1 containing only a single chain of PEG can be obtained if one of the included PEG chains is fortunately capped with a bulky molecule 3 during dissociation of the intermediary inclusion complex $2 \cdot \gamma$ -CD (Scheme 1). In such a loose-bound relationship between the components, y-CD would be endowed with additional inclusion of a guest polymer, as shown in Scheme 1. In this article, we report the preparation of a single-stranded polyrotaxane 1 composed of y-CD and a single PEG chain as a novel class of molecular assemblies with "loose" structure thanks to mismatching in terms of the size-adequacy, which was characterized by ¹H NMR and MALDI-TOF mass spectroscopy. It was demonstrated that the cavity of γ -CD in 1 provided accommodation for PEG guests (M_n : 400, 1000, and 2000) to form an additional inclusion complex through ¹H NMR and XRD analyses.

An inclusion complex of γ -CD and PEG-COOH **2** (M_n : 2000) was prepared by mixing both components in water^{4a,4b} as a precursor for the following end-capping reaction. Monoaminated β -CD derivative **3**⁷ was used as a bulky end-capping group to react with carboxylic terminals in the precursor complex. Single-stranded polyrotaxane **1** was isolated in 6% yield⁸ and has the shortest retention time in GPC in the reaction mixture. No rotaxanated species was found other than **1**, even with the assumption of diverse inclusion complexes between γ -CD and **2**, as drawn in Scheme 1. Instead, most of the PEG component was found in **4** (PEG-[β -CD-(OMe)₂₀]₂), which was produced in over 90% yield through dissociation, followed by condensation at the terminals of **2**. GPC profiles for the polyrotaxane **1** and its components, β -CD-appended PEG **4** and γ -CD, are shown in Figure S1 in the Supporting Information.

The structure of 1 was first investigated by 1H NMR spectroscopy in 1 wt 9 0 NaOD/D₂O. The spectrum was given as a simple

Scheme 1. Preparation of the Single-Stranded Polyrotaxane 1 through the End-Capping Reaction of Intermediary Inclusion Complex $2 \cdot \gamma$ -CD with 3, and Additional Inclusion Complexation of 1 with a PEG Guest



integration of γ -CD and β -CD-appended PEG **4** and indicated that the components were present in a ratio of ca. 10:1 to construct the polyrotaxane **1** (Figure S2 in the Supporting Information).

The MALDI-TOF mass spectrum of 1 is shown in Figure 1a. An envelope of peaks is centered at \sim 16 600, ranging from 11 200 to 21 000, in which a repeating distribution is separated from the

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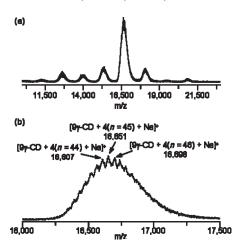


Figure 1. (a) MALDI-TOF mass spectrum of **1** and (b) its partial region around 16 600. Sinapinic acid was used as the matrix.

neighboring ones by 1297, corresponding to the molar mass of γ -CD. Detailed description of the MALDI-TOF mass spectrum is shown in Figure 1b. It was found that a repeating peak was separated from the neighboring peaks by 44, corresponding to the molar mass of an ethylene glycol (-CH₂CH₂O-) unit. The three peaks found at m/z values of 16698, 16651, and 16607, for instance, can be ascribed to members (n = 46, 45, 44) of singlestranded polyrotaxane 1 possessing nine γ -CD molecules (16 695, 16651, and 16607 are theoretically calculated values for these structures, respectively). The similar repeating set was also found in the other distributions (m = 5-12). Averaging these members can satisfy the 10:1 ratio in the ¹H NMR spectrum even without considering any double-stranded species, which is in accordance with the fact that there exist no distribution overlaps at the same regions in Figure 1. The single-stranded structure of 1 with two terminals was further supported by a Job plot based on the complexation-induced chemical shifts in ¹H NMR spectroscopy upon complexation of 1 with adamantane carboxylic acid in 1 M NaOD/D₂O, which is known to be a suitable guest molecule for binding with β -CD. The guest binding at the terminals of 1 was evidenced by a downfield shift of proton Ha in adamantane carboxylic acid. The 1:2 stoichiometry of the complex between 1 and adamantane carboxylic acid was confirmed by the plot exhibiting a peak at 0.67 (Figure 2).¹⁰

Considering the fact that the original γ -CD has a large enough cavity to include two molecules of PEG, the relation between γ -CD and the single PEG chain in 1 would be loose. Additional complexation of the single-stranded polyrotaxane 1 was examined with PEG chains (PEG400, PEG1000, and PEG2000) as a macromolecular guest. In the presence of the PEG guest, the polyrotaxane 1 was first dissolved in water adjusted to a pH of 13 and then neutralized. During the dissolution-neutralization process, the inclusion complex 1.PEG was formed as a white solid (Table S1 in the Supporting Information). The capturing of PEG into the solid was confirmed by the ¹H NMR spectrum showing the emphasized PEG signal. Powder X-ray diffraction analysis suggested the formation of inclusion complex of 1 with PEG₂₀₀₀ (Figure 3). A new peak at a 2θ of 7.5° was observed to indicate the emergence of a partially crystalline structure of γ -CD (Figure 3e), which is characteristic of γ -CD reported as the channel structure. ^{3e-3g,11} A similar peak is also observed in the diffractogram for the intermediary inclusion complex $2 \cdot \gamma$ -CD (Figure 3c). 4b-4e,11 These diffractograms are different from that of γ -CD (Figure 3a) or PEG-COOH 2 (Figure 3b) itself. It is noted that washing of the complex 1 · PEG with an excess amount of water led to the disappearance of both the emphasized PEG signal in the ¹H NMR spectrum and the peak at 7.5° in the XRD

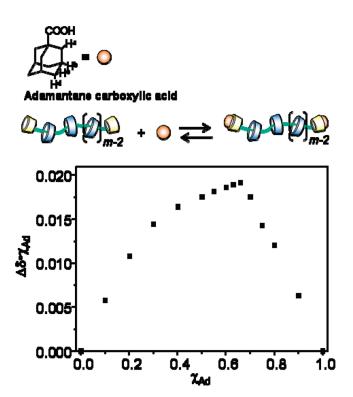


Figure 2. Job plot for the complexation of 1 with adamantane carboxylic acid (\blacksquare at proton H^a) in 1 M NaOD/D₂O at 303 K. χ_{Ad} denotes the molar fraction of adamantane carboxylic acid.

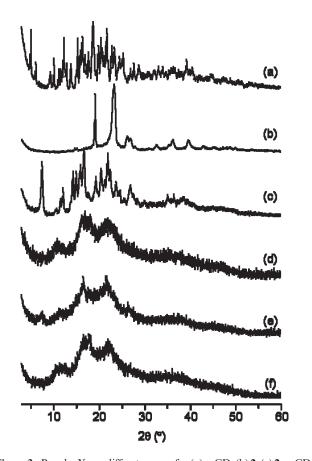


Figure 3. Powder X-ray diffractograms for (a) γ -CD, (b) **2**, (c) **2** · γ -CD, (d) **1**, (e) **1** · PEG₂₀₀₀, and (f) **1** · PEG₂₀₀₀ after washing with excess water.

diffractogram to give the identical spectrum and pattern (Figure 3f) to those of 1 itself (Figure 3d). Similar results from

the other combinations ($1 \cdot PEG_{400}$ and $1 \cdot PEG_{1000}$, Figure S3 in the Supporting Information) and a control experiment using α -CD-based polyrotaxane 5 are shown in Table S2 of the Supporting Information.

In conclusion, we have demonstrated the preparation of singlestranded "loose" polyrotaxane 1 composed of γ -CD and a single PEG chain. The MALDI-TOF mass spectrum revealed the single-stranded and doubly capped structure of 1 with repeating distributions of γ -CD and an ethylene glycol unit, which was further supported by a Job plot for the complexation at both terminals of 1 with adamantane carboxylic acid exhibiting the 1:2 stoichiometry. Complexation of 1 with additional PEG guests was demonstrated through the dissolution and neutralization process. In the complexes of 1.PEG, a crystalline structure of γ -CD was found. We are now studying the complexation properties of a water-solubilized "loose" polyrotaxane with various guests in solution. Also, such a macromolecular inclusion complexation can exploit possible future applications for smart materials such as carriers of macromolecular drugs, gelators in reversible polymer networks, and so on.

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Supporting Information Available: Experimental details (materials, analyses, and synthetic procedures), gel permeation chromatograms, complexation of 1 with PEG guests, X-ray diffractograms, complexation of 5 with PEG₂₀₀₀, ¹H NMR spectrum of 5, and Job plot for the complexation of 5 with adamantane carboxylic acid. This material is available free of charge via the Internet at http://pubs.acs.org.

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